## Photoelectron Spectra of Some Decahydrotrimethanoanthracenes: Observation of Large $\pi$ Orbital Interactions through Six Bonds and an Apparent Violation of the *trans* Rule

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Photoelectron spectroscopic studies of the trimethanoanthracenes have revealed a large  $\pi$  orbital interaction through six bonds, an apparent violation of the *trans* rule and a new and effective way by which a CH<sub>2</sub> group can act as a through-bond relay.

Calculations have led to the following two important predictions concerning the dependence of orbital interactions through bonds<sup>1</sup> (OITB) on the conformation and length of the relaying  $\sigma$  framework. (i) OITB are maximized for an all-

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 $= CH_2 - CH_2; Y - Y = CH = CH$ 

**Table 1.** Vertical  $\pi I_p$ 's (eV) and PRDDO assignments and splitting energies  $\Delta E$ (PRDDO).

Compound	$\pi~I_{p}$	Assignment	$\Delta I_{ m p}$	$\Delta E(PRDDO) /eV$
(1)	8.58	$\pi - \pi$	0.32	0.17
	8.90	$\pi + \pi$		
(2a)	8.36	$\pi - \pi$	0.38	0.23
	8.74	$\pi + \pi$		
(2b)	8.68	π		
(2c)	8.47	$\pi$		
(3a)	8.24	$\pi - \pi$	0.52	0.49
	8.76	$\pi + \pi$		
( <b>3b</b> )	8.51	$\pi$		
(4)	8.48ª	$\pi - \pi$	0.87	0.75
	9.35ª	$\pi + \pi$		
Ref. 3.				

*trans* (*i.e.* staggered) arrangement of  $\sigma$  bonds (the *trans* rule).<sup>1-4</sup> (ii) Interactions through n bonds (OIT-n-B) are attenuated only mildly with increasing n such that OIT-6-B and OIT-8-B could be as large as 0.2 and 0.1 eV, respectively.<sup>1,3,4</sup> The exiguity of experimental data has made the generality of these predictions difficult to assess. Thus only three studies, two on OIT-4-B<sup>3,5</sup> and one on OIT-5-B,<sup>6</sup> have provided supportive evidence for the trans rule. From a photoelectron study of a series of di-iodopolyethynes it seemed that OIT-n-B are attenuated with increasing n more rapidly than predicted, being 0.3 eV for OIT-5-B and negligible for OIT-7-B.7 We now report the results of the first experimental study of OIT-6-**B** which reveal not only that such interactions can be large but also an apparent violation of the trans rule.

The He I photoelectron spectra of the trimethanoanthracenes (1),<sup>8</sup> (2),<sup>8</sup> and (3),<sup>8,9</sup> have been measured and the vertical  $\pi I_p$ 's are given in Table 1.

The  $\pi \Delta I_{\rm p}$  of 0.32 eV for (1) is a direct measure of  $\pi OIT$ -6-B in this compound and is the largest value so far reported for orbitals interacting through more than 4 bonds. Indeed this result, together with that observed for OIT-4-B in (4)<sup>3</sup> (Table



1), suggests that OIT-8-B in the next higher 'norbornylogue' of (1) i.e. the corresponding tetramethanonaphthacene, should be 0.1-0.15 eV and therefore measurable. PRDDO MO<sup>10</sup> (PRDDO = partial retention of diatomic differential overlap) calculations on (1) gave the level ordering  $\pi - \pi > \pi + \pi$  in agreement with the parity rule.

Surprisingly the  $\pi \Delta I_p$ 's increase along the series (1), (2a), (3a), reaching a maximum of 0.52 eV for (3a). This trend, which is also qualitatively reproduced by PRDDO calculations, violates the trans rule since on going from (1) to (3a) the number of *trans* arrangements of vicinal  $\sigma$  bonds progressively decreases. However this violation is only apparent and is due to the presence of additional orbital interactions mediated by the central bridging  $CH_2$  group in (3a) and, to a lesser extent, in (2a) but which are virtually absent in (1). Supportive of this view is the observation that the measured  $\pi \Delta I_{\rm p}$  of (5), whose geometry is expected to resemble that of (3a) minus the central CH<sub>2</sub> group, is only ca. 0.2 eV.<sup>11</sup> This value may be taken as a measure of the magnitude of OIT-6-B in both (5) and (3a). It then follows that interactions involving the CH<sub>2</sub> group in (3a) amount to 0.32 eV (= 0.52 - 0.20 eV). A direct estimate of the magnitude of the mediating effect of the CH<sub>2</sub> group on the  $\pi$  interactions in (3a) was obtained from an STO-3G ab initio calculation on a model ethene-methaneethene system, (6), whose configuration mimics that expected of the corresponding groups in (3a).<sup>†</sup> The calculated  $\pi,\pi$ splitting energy of 0.38 eV (which disappeared when the CH<sub>4</sub> molecule was removed) is in good agreement with the value of 0.32 eV obtained above by indirect means. A similar calculation on the ethene-methane-ethene system whose configuration now mimics that of the corresponding groups in (2a) gave a splitting energy of 0.1 eV. Thus OIT-6-B in (2a) are estimated to be 0.28 eV, i.e. 0.1 eV larger than those predicted for (3a). Therefore the trans rule is satisfied for the series (1), (2a), and (3a) provided that OIT-6-B alone are considered.

The lowering of the  $\pi$  I<sub>p</sub>'s of (2c) and (3b) by, respectively, 0.21 and 0.17 eV compared to (2b) is due to interactions between the central CH<sub>2</sub> group and the double bond in the former two molecules. PMO theory then tells us that the mixing of the CH<sub>2</sub> group with two double bonds, as in (3a), should be double the above interaction *i.e.* 0.34-0.42 eV. This value is in excellent agreement with that calculated for (6) (vide supra).

<sup>†</sup> The geometry was based on the known structure of aldrin in which the CH<sub>2</sub>-double bond distance is 3.1 Å: T. P. DeLacy and C. H. L. Kennard, J. Chem. Soc., Perkin Trans. 2, 1972, 2153.

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## References

- (a) R. Hoffmann, A. Imamura, and W. J. Hehre, J. Am. Chem. Soc., 1968, 90, 1499; (b) R. Hoffmann, Acc. Chem. Res., 1971, 4, 1.
- 2 T. K. Brunck and F. Weinhold, J. Am. Chem. Soc., 1976, 98, 4392.
- 3 M. N. Paddon-Row, H. K. Patney, R. S. Brown, and K. N. Houk, J. Am. Chem. Soc., 1981, 103, 5575.

- 4 M. N. Paddon-Row, Acc. Chem. Res., 1982, 15, 245.
- 5 V. Balaji, K. D. Jordan, P. D. Burrow, M. N. Paddon-Row, and H. K. Patney, J. Am. Chem. Soc., 1982, 104, 6849.
- 6 P. Pasman, J. W. Verhoeven, and Th. J. de Boer, Tetrahedron Lett., 1977, 207.
- 7 E. Heilbronner, V. Hornung, and E. Kloster-Jensen, *Helv. Chim. Acta*, 1970, 53, 331; E. Heilbronner, V. Hornung, J. P. Maier, and E. Kloster-Jensen, *J. Am. Chem. Soc.*, 1974, 96, 4252; G. Bieri, E. Heilbronner, T. B. Jones, E. Kloster-Jensen, and J. P. Maier, *Phys. Scr.*, 1977, 16, 202.
- 8 M. N. Paddon-Row and H. K. Patney, Aust. J. Chem., submitted for publication.
- 9 J. Mantzaris and E. Weissberger, J. Org. Chem., 1974, 39, 726.
- 10 T. A. Halgren, D. A. Kleier, J. H. Hall, Jr., L. D. Brown, and
- W. N. Lipscomb, J. Am. Chem. Soc., 1978, 100, 6595. 11 G. D. Willett, personal communication to M.N. P.-R.